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APPLICATION NO. FILING DATE FIRST NAMED INVENTOR ATTORNEY DOCKET NO. CONFIRMATION NO. 09/664.332 09/18/2000 Noriya Hayashi 001195 4422 23850 01/31/2005 EXAMINER ARMSTRONG, KRATZ, QUINTOS, HANSON & BROOKS, LLP SELLERS, ROBERT E 1725 K STREET, NW ART UNIT PAPER NUMBER **SUITE 1000** WASHINGTON, DC 20006 1712

DATE MAILED: 01/31/2005

Please find below and/or attached an Office communication concerning this application or proceeding.

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		Applic	ation No.	Applicant(s)	
		09/664	1,332	HAYASHI, NORIYA	
	Office Action Summary	Exami	ner	Art Unit	
			Sellers	1712	
Period f	The MAILING DATE of this communic or Reply	ation appears on	the cover sheet wi	th the correspondence ad	dress
THE - External and afternal files af	MORTENED STATUTORY PERIOD FOR MAILING DATE OF THIS COMMUNIC ensions of time may be available under the provisions of SIX (6) MONTHS from the mailing date of this commuse period for reply specified above is less than thirty (30) of period for reply is specified above, the maximum stature to reply within the set or extended period for reply we reply received by the Office later than three months after the patent term adjustment. See 37 CFR 1.704(b).	CATION.  f 37 CFR 1.136(a). In no nication.  days, a reply within the substray period will apply and ill, by statute, cause the	event, however, may a r statutory minimum of third d will expire SIX (6) MON application to become AB	eply be timely filed  y (30) days will be considered timely THS from the mailing date of this co	
Status					
1)⊠	Responsive to communication(s) filed on <u>27 December 2004</u> .				
·	•	o)⊠ This action is			
3)□	Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under <i>Ex parte Quayle</i> , 1935 C.D. 11, 453 O.G. 213.				
Disposit	ion of Claims				
5)□ 6)⊠ 7)□	Claim(s) 1-3,6-10,12,17-19 and 21-28 is/are pending in the application.  4a) Of the above claim(s) 9,17-19,21 and 23-26 is/are withdrawn from consideration.  Claim(s) is/are allowed.  Claim(s) 1-3, 6-8, 10, 12, 22, 27 and 28 is/are rejected.  Claim(s) is/are objected to.  Claim(s) are subject to restriction and/or election requirement.				
Applicat	ion Papers				
10)	The specification is objected to by the The drawing(s) filed on is/are: Applicant may not request that any object Replacement drawing sheet(s) including to The oath or declaration is objected to	a)⊡ accepted or ion to the drawing(s he correction is req	s) be held in abeyan uired if the drawing(	ce. See 37 CFR 1.85(a). (s) is objected to. See 37 CF	
Priority (	under 35 U.S.C. § 119				
12)□ a)	Acknowledgment is made of a claim for All b) Some * c) None of:  1. Certified copies of the priority d  2. Certified copies of the priority d  3. Copies of the certified copies of application from the Internation.  See the attached detailed Office action	ocuments have b ocuments have b the priority docu al Bureau (PCT R	een received. een received in A ments have been Rule 17.2(a)).	pplication No received in this National	Stage
Attachmen	, t(s)				
	ee of References Cited (PTO-892)	0.040)		ummary (PTO-413)	
3) 🔲 Infor	e of Draftsperson's Patent Drawing Review (PTo mation Disclosure Statement(s) (PTO-1449 or P or No(s)/Mail Date			)/Mail Date formal Patent Application (PTO 	<b>-152</b> )

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1. This is responsive to the Request for Continued Examination filed December 27, 2004 along with the 37 CFR 1.132 declaration.

2. Claims 9, 17-19, 21 and 23-26 are withdrawn from further consideration pursuant to 37 CFR 1.142(b) as being drawn to nonelected species, there being no allowable generic or linking claim. Election was made **without** traverse in the reply filed on March 6, 2002.

The text of section 103(a) of Title 35, U.S. Code not included in this action can be found in a prior Office action.

Claims 1-3, 6-8, 10, 12, 22, 27 and 28 are rejected under 35 U.S.C. 103(a) as being unpatentable over Hamazu et al. Patent No. 5,359,017; Buchwalter et al. Patent No. 5,879,859; Starkey Patent No. 5,384,339 and Green Patent No. 4,252,592 in view of Green Patent No. 4,299,938.

The rejection is maintained for the reasons of record set forth in the previous Office actions. The arguments and declaration filed December 24, 2004 have been considered but are unpersuasive.

- 3. The declaration attempts to demonstrate the criticality of the claimed molar ratio of anhydride curing agent to photopolymerizable resin of from 0.3:1 to 1.4:1 by comparing the extent of curing of Test I (molar ratio of 0.65:1) with Test II (molar ratio of 0), Test III (molar ratio of 0.01:1) and Test IV (molar ratio of 2.5:1).
- 4. Hamazu et al. does not exemplify compositions with an anhydride curing agent, although an acid anhydride is disclosed in column 5, line 14.

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5. Green '592 does not exemplify compositions containing an anhydride curing agent, although an acid anhydride is disclosed in column 4, line 17.

- 6. Starkey does not exemplify formulations with an anhydride hardener, although the use of an anhydride thermohardening catalyst (col. 21, lines 33-49) such as the elected species of maleic anhydride (col. 21, lines 38-39) is disclosed at a level of as much as 10 parts by weight per 100 parts by weight of an epoxy resin (col. 21, lines 12-15). The elected species of 3,4-epoxycyclohexylmethyl-3,4-epoxycyclohexane carboxylate is shown in Examples 2 (cols. 21-22) and 5 (col. 23) along with an aromatic sulfonium salt. The molar quantity of maleic anhydride is  $10 \div 98.6$  g/mole = 0.316 mole. The molar concentration of 3,4-epoxycyclohexylmethyl-3,4-epoxycyclohexane carboxylate is  $100 \div 316 = 0.316$  mole. The molar ratio of curing agent:photopolymerizable resin is  $0.10 \div 0.316 = 0.32:1$  which is within the claimed parameters.
- 7. Buchwalter et al. (col. 9, Example 1) shows a blend of 1.8 parts by weight of acetal diepoxide (i.e. acetaldehyde bis(3,4-epoxycyclohexylmethyl)acetal according to col. 3, lines 13-15), 0.91 part by weight of hexahydrophthalic anhydride and a photoinitiator such as the disclosed sulfonium salt (col. 3, line 22). The molar amount of anhydride is 0.91 ÷ 154.17 g/mole = 0.0059 mole. The molar proportion of diepoxide is 1.8 ÷ 283 g/mole = 0.0064 mole. The molar ratio of curing agent:photopolymerizable resin is 0.0059 ÷ 0.0064 = 0.93:1.

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8. Accordingly, the closest prior art example is Example 1 of Buchwalter et al. wherein the molar ratio of curing agent:photopolymerizable resin is 0.93:1 which is embraced by the claimed limits of from 0.3:1 to 1.4:1. Tests II-IV of the declaration are not germane to the composition of Buchwalter et al. since the molar ratio is exhibited.

- 9. The references set forth epoxy resins which are within photopolymerizable resin according to page 26, lines 9-11 of the instant specification, aromatic sulfonium salts such as the elected species of benzyl-4-hydroxyphenylmethylsulfonium hexafluoroantimonate of Hamazu et al. (col. 3, lines 29-30, corresponding to claimed formula IV) and anhydride curing agents with a molar ratio of curing agent:photopolymerizable resin of 0.32:1 disclosed in Starkey or 0.93:1 shown in Buchwalter et al.
- 10. The claims are directed to a composition "which can be cured with an energy ray (claim 1, line 2)" and "which makes it possible to cure by chain reaction (line 3)." There are no affirmative limitations requiring the curing of the composition by energy ray and by chain reaction. The prior art considered as a whole shows (with respect to Buchwalter et al.) or discloses (regarding Hamazu et al., Starkey and Green '592) the claimed mixture of epoxy resin, anhydride curing agent and an aromatic sulfonium salt which is within the realm of the claims regardless of the possible curing mechanism.

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11. Even if, arguendo, the curing mechanism is considered, the capability of curing the compositons of the references by energy ray addresses the claimed capability of curing by irraditation with an energy ray which embraces ultraviolet radiation (specification, pages 66-67, Example 2). Hamazu et al. (col. 22, lines 67-68 and col. 23, line 61 to col. 24, line 3, ultraviolet radiation), Buchwalter et al. (col. 9, lines 14-15, electromagnetic radiation), Starkey (col. 14, lines 3-5, ultraviolet radiation) and Green '592 (col. 2, lines 7-15, actinic radiation) espouse curing by energy ray.

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- 12. Based on the equivalent formulations of the prior art and claims containing an epoxy photopolymerizable resin, an anhydride curing agent and an aromatic sulfonium salt encompassed by formula IV (Hamazu et al.) curable by energy ray, the compositions of Hamazu et al., Buchwalter et al., Starkey and Green '592 inherently cure by chain reaction.
- 13. Green '938 teaches a polyhydric alcohol such as polyethylene glycol (col. 8, lines 13-14) as a co-curing agent with an anhydride (col. 12, lines 18-19) as required in claims 2, 6, 7 and 10 in a formulation comprising 3,4-epoxycycloyhexylmethyl-3,4-epoxycyclohexane carboxylate (col. 7, lines 54-55) and an aryoxysulfoxonium salt photoinitiator (col. 4, line 55 to col. 5, line 12). Starkey espouses a mixture of an anhydride and a polyol such as polyethylene glycol (col. 7, line 31) which adjusts the molecular weight and controls the degree of

crosslinking (col. 3, lines 17-19; col. 7, line 67-68 and col. 8, lines 14-15).

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14. It would have been obvious to incorporate the polyethylene glycol of Green '938 and Starkey with the anhydride curing agent of Hamazu et al. and Buchwalter et al. in order to control the degree of crosslinking (Starkey).

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Robert Sellers Primary Examiner Art Unit 1712